

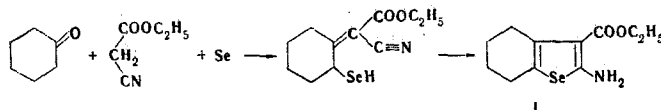
SYNTHESIS OF 2-AMINO-3-CARBETHOXY-4,5,6,7-TETRAHYDROBENZO[b]SELENOPHENE AND SOME TRANSFORMATIONS BASED ON IT

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UDC 547.739.3.07

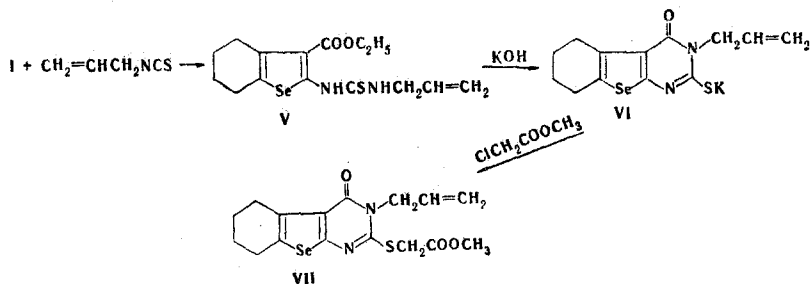
A new method for the synthesis of 2-amino-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]selenophene was developed. The reaction of the latter with allyl isothiocyanate gave 2-(N'-allylthioureido)-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]selenophene, which is cyclized to the potassium salt of 3-allyl-4-oxo-2-thio-3,4,5,6,7,8-hexahydrobenzo[b]selenopheno[2,3-d]pyrimidine on treatment with potassium hydroxide.

Up until now, the most convenient method for the preparation of benzo[b]selenophene has been a multistep synthesis from anthranilic acid and sodium diselenide [1]. Methods for the preparation of 2-amino-selenophene derivatives on the basis of the reaction of elemental selenium with cyanoacetic ester and ketones, which is a method similar to that used to obtain 2-aminothiophene derivatives [2], are not presently known. Cyanoacetic ester and ketones do not react with selenium in ethanol and in higher-boiling solvents [dimethyl sulfoxide (DMSO) and dimethylformamide (DMF)] in the presence of diethylamine. 2-Amino-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]selenophene (I) could be obtained only on prolonged heating with dimethylacetamide as the solvent.



The amino group of I readily reacts with maleic and succinic anhydrides and 2,4-dinitrobenzaldehyde to give, respectively, the monoamides of maleic (II) and succinic (III) acids and azomethine derivative (IV).

In order to obtain a condensed system from selenophene and pyrimidine rings, we synthesized 2-(N'-allylthioureido)-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]selenophene (V) and obtained the potassium salt of 3-allyl-4-oxo-2-thio-3,4,5,6,7,8-hexahydrobenzo[b]selenopheno[2,3-d]pyrimidine (VI) by the action of potassium hydroxide on it. The salt readily reacts with methyl chloroacetate to give 3-allyl-2-(carbo-methoxymethylthio)-4-oxo-3,4,5,6,7,8-hexahydrobenzo[b]selenopheno[2,3-d]pyrimidine (VII).



Uzhgorod State University. Translated from *Khimiya Geterotsiklicheskikh Soedinenii*, No. 3, pp. 326-328, March, 1973. Original article submitted February 21, 1972.

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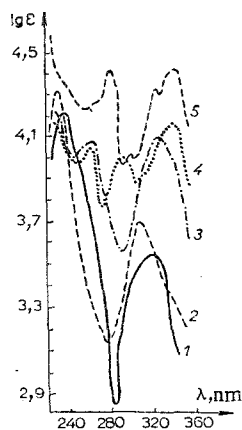


Fig. 1. UV spectra: 1) 2-amino-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]selenophene (I) 2) 2-amino-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]thiophene (VIII); 3) 3-allyl-2-(carbomethoxymethylthio)-4-oxo-3,4,5,6,7,8-hexahydrobenzo[b]selenopheno[2,3-d]pyrimidine (VII); 4) potassium salt of 3-allyl-4-oxo-2-thio-3,4,5,6,7,8-hexahydrobenzo[b]selenopheno[2,3-d]pyrimidine (VI), 5) potassium salt of 3-allyl-4-oxo-2-thio-3,4,5,6,7,8-hexahydrobenzo[b]thieno[2,3-d]pyrimidine (IX).

The UV spectra of I and VI are similar to the spectra of the corresponding sulfur-containing derivatives - 2-amino-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]thiophene (VIII) [3] and the potassium salt of 3-allyl-4-oxo-2-thio-3,4,5,6,7,8-hexahydrobenzo[b]thieno[2,3-d]pyrimidine (IX) (see Fig. 1).

The absorption maxima of selenium-containing compounds I and VI are shifted as compared with the maxima of the corresponding sulfur-containing analogs (VIII and IX) and have lower extinctions; this is probably explained by the fact that the electron pair of selenium in the selenium compounds participates to a lesser extent in conjugation of the sextet of electrons of the selenophene ring than the electron pair of sulfur in the thiophene ring of VIII and IX.

## EXPERIMENTAL

The UV spectra of  $10^{-5}$  M solutions of the compounds in ethanol (d 1 cm) were recorded with an SF-4A spectrometer. The IR spectrum of I was recorded with a UR-10 spectrophotometer.

2-Amino-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]selenophene (I). A 7.89-g (0.1 g-atom) sample of selenium, 2.94 g (0.3 mole) of cyclohexanone, 33.9 g (0.3 mole) of ethyl cyanoacetate, 5 ml of diethylamine, and 50 ml of dimethylacetamide were refluxed for 25 h, after which the hot solution was filtered, and 2 volumes of water were added to the filtrate. The resulting oily liquid began to crystallize in the cold after 24 h to give 2.2 g (8%) of a product with mp 137-138° (from ethanol).  $\lambda_{\max}$ , nm (log  $\epsilon$ ): 235 (4.21), 320 (3.55). IR spectrum (KBr pellet): 3400, 3290  $\text{cm}^{-1}$  ( $\nu_{\text{NH}}$ ), 1640, 1580  $\text{cm}^{-1}$  ( $\delta_{\text{NH}_2}$ ), 1300, 1050  $\text{cm}^{-1}$  ( $\nu_{\text{COC}}$ ). Found: C 48.6; 48.6; N 5.3; 5.2; Se 29.2; 29.3%.  $\text{C}_{11}\text{H}_{15}\text{NO}_2\text{Se}$ . Calculated: C 48.5; N 5.2; Se 29.0%.

2-(3-Carboxyacryloylamino)-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]selenophene (II). A thoroughly ground mixture of 0.27 g (0.001 mole) of I and 0.098 g (0.001 mole) of maleic anhydride was fused at 100°. The mixture melted and immediately began to solidify to give 0.36 g (98%) of a product with mp 172-173° (from ethanol).  $\lambda_{\max}$ , nm (log  $\epsilon$ ): 285 (3.98), 340 (4.02). Found: C 48.8; 48.6; N 3.9; 4.0; Se 21.6; 21.7%.  $\text{C}_{15}\text{H}_{17}\text{NO}_5\text{Se}$ . Calculated: C 48.6; N 3.8; Se 21.3%.

2-(3-Carboxypropionamido)-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]selenophene (III). A mixture of 0.27 g (0.001 mole) of I and 0.1 g (0.001 mole) of succinic anhydride was heated in dry benzene for 1 h, after which it was cooled to precipitate 0.34 g (92%) of a product with mp 160-162° (from methanol).  $\lambda_{\max}$ , nm (log  $\epsilon$ ): 275 (3.70), 282 (3.55), 315 (3.76), 325 (3.80). Found: C 48.6; 48.4; N 3.8; 3.8%.  $\text{C}_{15}\text{H}_{19}\text{NO}_5\text{Se}$ . Calculated: C 48.4; N 3.8%.

2-(2,4-Dinitrobenzylideneamino)-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]selenophene (IV). A mixture of 0.14 g (0.0005 mole) of I and 0.098 g (0.0005 mole) of 2,4-dinitrobenzaldehyde was pulverized thoroughly and fused at 100°. The melt began to solidify on cooling to give 0.21 g (95%) of a product with mp 59-61° (from ethanol).  $\lambda_{\max}$ , nm (log  $\epsilon$ ): 240 (4.59), 305 (3.76). Found: C 48.1; 48.2; N 9.4; 9.4%.  $\text{C}_{18}\text{H}_{17}\text{N}_3\text{O}_6\text{Se}$ . Calculated: C 48.0; N 9.3%.

2-(N<sup>1</sup>-Allylthioureido)-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]selenophene (V). A mixture of 0.95 g (0.0035 mole) of I and 0.35 g (0.0035 mole) of allyl isothiocyanate was heated in 7 ml of ethanol for 7 h on a boiling-water bath, after which it was cooled to give 1.3 g (90%) of V with mp 131-133° (from methanol).  $\lambda_{\max}$ , nm (log  $\epsilon$ ): 235 (3.56), 242 (3.50), 285 (3.47), 295 (3.53), 318 (3.42), 340 (3.70). Found: C 48.5; 48.6; N 7.4; 7.6%.  $\text{C}_{15}\text{H}_{20}\text{N}_2\text{O}_2\text{SSe}$ . Calculated: C 48.5; N 7.6%.

Potassium Salt of 3-Allyl-4-oxo-2-thio-3,4,5,6,7,8-hexahydrobenzo[b]selenopheno[2,3-d]pyrimidine (VI). A solution of 0.37 g (0.001 mole) of V in 5 ml of ethanol and 2 ml of a 2 N alcohol solution of potassium hydroxide was heated on a boiling-water bath for 30 min, after which it was cooled to give 0.28 g

(77%) of VI with mp 280° (dec.).  $\lambda_{\max}$ , nm (log  $\epsilon$ ): 260 (4.05), 295 (4.00), 340 (4.18). Found: C 43.0; 42.8; N 7.8; 7.7%.  $C_{13}H_{13}KN_2OSse$ . Calculated: C 43.0; N 7.7%.

3-Allyl-2-(carbomethoxymethylthio)-4-oxo-3,4,5,6,7,8-hexahydrobenzo[b]selenopheno[2,3-d]pyrimidine (VII). A mixture of 0.36 g (0.001 mole) of VI and 0.11 g (0.001 mole) of methyl chloroacetate in 5 ml of methanol was heated at 100° for 2 h, after which it was cooled to give a flocculent precipitate, which was removed by filtration and washed with water to give 0.36 g (91%) of a product with mp 103-105° (from ethanol).  $\lambda_{\max}$ , nm (log  $\epsilon$ ): 265 (4.07), 330 (4.10). Found: C 48.4; 48.4; N 7.0; 7.0%.  $C_{16}H_{18}N_2O_3Sse$ . Calculated: C 48.4; N 7.1%.

2-Amino-3-carbethoxy-4,5,6,7-tetrahydrobenzo[b]thiophene (VIII) [3].  $\lambda_{\max}$ , nm (log  $\epsilon$ ): 230 (4.31), 308 (3.70).

Potassium Salt of 3-Allyl-4-oxo-2-thio-3,4,5,6,7,8-hexahydrobenzo[b]thieno[2,3-d]pyrimidine (IX) [4].  $\lambda_{\max}$ , nm (log  $\epsilon$ ): 285 (4.42), 305 (4.02), 330 (4.32), 350 (4.41).

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